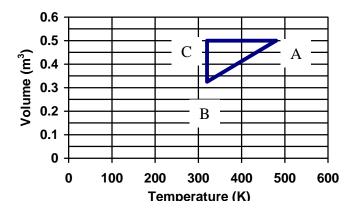
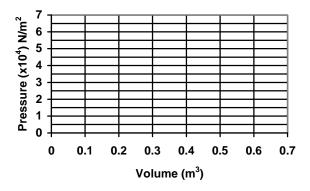
8 moles of ideal diatomic gas are used in a process that goes from A to B to C. T<sub>A</sub>=480K, T<sub>B</sub>=320K. Complete the p-V diagram on the right.





Complete the charts for Process A to B.

P

Q

 $U_{\rm f}$ 

I F I F I F

Complete the charts for Process B to C.

P

V

T

 $U_{i}$ 

Q

W

 $U_{\rm f}$ 

ΙF I F ΙF

Complete the charts for Process C to A.

P

V

T

 $U_{i}$ 

Q

W

 $U_{\rm f}$ 

I F I F I F

What kind of process is C to A?

What was the entropy change of the gas from B to C?\_\_\_\_\_